

Energy

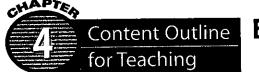
Underlined words and phrases are to be filled in by students on the Note-taking Worksheet.

The Nature of Energy Section 1

- A. Energy is the ability to cause change.
 - 1. Kinetic energy—energy in the form of motion
 - a. The amount of kinetic energy an object has depends on its mass and its speed.
 - **b.** Kinetic energy = $\frac{1}{2}$ mass × speed²
 - c. Joule—the SI unit used to measure energy
 - 2. Potential energy—energy stored in a motionless object, giving it the potential to cause change
 - 3. Elastic potential energy—energy stored by things that stretch or compress
 - 4. Chemical potential energy—energy stored in chemical bonds between atoms
 - 5. Gravitational potential energy—energy stored by things that are above Earth
 - a. The amount of GPE an object has depends on its mass, the acceleration due to gravity, and its height above ground.
 - **b.** GPE = mass in kilograms \times 9.8 m/s² \times height in meters

Discussion Question

How does food provide energy for your body? Chemical potential energy is stored in the chemical bonds between atoms in the food. When you digest the food, your body breaks down the molecules in the food, breaking the chemical bonds and releasing energy.



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Section 2 Conservation of Energy

- A. Energy conversions—energy changing from one form to another
 - 1. Fuels store energy in the form of chemical potential energy.
 - 2. <u>Mechanical</u> energy—the total amount of potential and kinetic energy in a system
- **B.** Law of Conservation of Energy—energy may change from one form to another, but the total amount of energy never changes.
 - 1. Example—as a swing moves back and forth, its energy continually converts from kinetic to potential and back.
 - 2. If the energy of the swing decreases, then the energy of some other object must <u>increase</u> by an equal amount.
 - 3. Friction converts some of the mechanical energy into thermal energy.
- C. Converting <u>mass</u> into energy—you must think of mass as energy when discussing nuclear reactions. The total amount of mass and energy is conserved.
 - 1. Nuclear fusion—two nuclei are fused together—takes place in the sun.
 - 2. Nuclear fission—two nuclei are broken apart.
- D. Conservation of energy in your body
 - 1. <u>Chemical potential</u> energy from food that is stored in your body is used to fuel the processes that keep you alive.
 - 2. The food <u>calorie</u> is used to measure how much energy you get from various foods. One Calorie is equivalent to about 4,180 J.

Discussion Question_

Is energy lost when a car burns gasoline? Explain. No, energy is always conserved. When a car burns gasoline, that energy is changed to kinetic energy, which makes the car move. In turn, friction from the tires on the road converts some of that kinetic energy into thermal energy, which remains in the road and the tires.

END